## VP & RPTP Science College-Vallabh Vidyanagar

Internal Test: 2016

## S Y BSc [Semester-IV]

## Subject-Physics US04CPHY02

Date:	15/0	3/2016 Tuesday	Time: 03.00 p	m to 0	4.30 pm	Total Marks-25	
Q-1	Mult	Multiple Choice Questions: [Attempt all]					
(i)	The	The number of atoms present in unit cell of HCP structure is					
	(a)	1		(b)	2	Science	
	(c)	4		(d)	6	TRAPY :	
(ii)	has a covalent bond.					IBKAK!	
	(a)	Na		(b)	Mg *	V. V. N893	
	(c)	Cu		(d)	Si		
(iii)	If all	If all the hydrogen atoms in polyethylene are replaced by fluorine, the resulting					
	poly	mer is called					
	(a)	Polyvinyl chlorid	e .	(b)	Polypropylene		
	(c)	Polytetrafluoroet	thylene	(d)	Polycarbonate		
Q-2	Ans	swer the following	g questions in sh	nort [A	attempt any two].	4	
(	a) Def	Define primitive cell and non-primitive cell.					
(	b) Wh	ich forces are negle	ected in cohesion	? Why	?		
(	c) Giv	e two differences b	etween thermop	lastic a	and thermosetting p	olymers.	
Q-3	Def	ine atomic packing	factor (APF). Cal	culate	APF for BCC structu	re. 6	
		OR					
Q-3 Describe the scheme to determine the Miller indices of a plane						d draw 6	
	pla	planes having Miller indices (1 1 0) and (1 1 1).					
Q-4	Cal	Calculate the lattice energy of ionic crystals.					
		OR					
Q-4	Derive the Madelung constant of ionic crystals.					6	
Q-5	Give classification of polymers based on molecular characteristics also					also 6	
	disc	discuss stereo-isomerism in detail.					
			OH	R			
Q-5	Wr	Write a note on: (1) Elastomers (2) Fibers					