



V. P. & R. P. T. P. Science College

B.Sc. (sixth semester)

Tuesday 13<sup>th</sup> March 2018

Subject code: US06CICV02 (Heavy and Fine Organic Chemistry)

Industrial Chemistry

Time: 11:00 am to 12:30 pm

Total Marks: 30

**Q-1 Select right option from given in the following questions.**

Marks-03

I. Vapour phase reaction between acetylene and acetic acid in the presence of a \_\_\_\_\_ catalyst yields vinyl acetate.

(a) Mercuric chloride

(b) Zinc acetate

(c) Charcoal

(d) Raney-nickel

II. The catalytic hydrogenation of dextrose yields \_\_\_\_\_

(a) Acetone

(b) Glucose

(c) Sorbitol

(d) Ethanol

II. In manufacture of methyl chloride, reaction gases pass through a water scrubber to remove \_\_\_\_\_

(a) H<sub>2</sub>S

(b) CO

(c) HCl

(d) NO<sub>2</sub>

**Q-2 Answer any two of the following:**

Marks-04

i. Write uses of Vinyl acetate

ii. Draw the flow diagram of manufacturing formaldehyde from methanol

iii. Write uses of Chloroform

Q-3 Write a manufacturing process of Phenol by cumene process

Marks-06

OR

Q-3 Vinyl chloride from acetylene and hydrogen chloride

Marks-06

Q-4 Write a note on triphenyl phosphine

Marks-06

OR

Q-4 write a manufacturing process of Formaldehyde

Marks-06

Q-5 write a manufacturing process of Carbon tetrachloride with flow diagram

Marks-06

OR

Q-5 Write a note on dialkylaminoethanol (DEA)

Marks-06